

Science 9 Measurement in the Lab

Name: $\qquad$
Date: $\qquad$
Block: $\qquad$
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Throughout the course we will be collecting information (data) from measurements that we make. It is important to have an idea of how to make these measurements so today we will practice!

| EQUIPMENT | NOTES |
| :---: | :---: |
|  | How to use: <br> Units used: |
|  | How to use: <br> Units used: |
|  | How to use: <br> Units used: |
| $\underbrace{35}$ | How to use: <br> Units used: |

## Measuring the Volume of Solid Objects

Let's figure out how to measure the volume of a regularly shaped object and an irregularly shaped object.

| REGULAR SHAPE | IRREGULAR SHAPE |
| :---: | :---: |

## Measuring the Volume Liquids

When reading and recording $L$ $\qquad$ measurements, we will be using a g $\qquad$ (measuring) c $\qquad$ .

When doing so, we MUST read from the $\qquad$ .

The meniscus is $\qquad$


Measurement Activity - Using the equipment around the room complete the tables below:

## A. Regular Shaped Objects

## Measuring Mass

| Object | Estimated mass (g) | Measured mass (g) |
| :--- | :--- | :--- |
|  |  |  |
|  |  |  |
|  |  |  |

Measuring and Calculating Regular Volume (length x width x height)

| Object | Estimated <br> volume $\left(\mathrm{cm}^{3}\right)$ | Measured <br> length $(\mathrm{cm})$ | Measured <br> Width (cm) | Measured <br> Height (cm) | Volume (cm $\left.{ }^{3}\right)$ |
| :--- | :--- | :--- | :--- | :--- | :--- |
|  |  |  |  |  |  |
|  |  |  |  |  |  |
|  |  |  |  |  |  |

B. Irregular Shaped Objects

## Measuring Mass

| Object | Estimated mass (g) | Measured mass (g) |
| :--- | :--- | :--- |
|  |  |  |
|  |  |  |
|  |  |  |

## Measure Irregular Volume

| Object | Estimated <br> volume $(\mathrm{mL})$ | Initial Volume <br> $(\mathrm{mL}$ w/out object $)$ | Final Volume <br> $(\mathrm{ml}$ with object $)$ | Actual Volume <br> $(\mathrm{mL}$ final -mL initial $)$ |
| :--- | :--- | :--- | :--- | :--- |
|  |  |  |  |  |
|  |  |  |  |  |
|  |  |  |  |  |

## C. Measuring Temperature

## Procedure:

1. Make an estimate of the temperature $\left({ }^{\circ} \mathrm{C}\right)$ for both the tap water and the water with ice cubes and salt.
2. Using the graduated cylinder measure out 75 mL of tap water.
3. Pour the water from the graduated cylinder into a beaker.
4. Use the thermometer to measure the temperature. Be sure to grasp the
 thermometer from the top and hold the thermometer in the middle of the solution (see picture on the left). DO NOT let the thermometer touch the bottom or the side of the beaker.
5. Dump out the water in to the sink.
6. Measure out 75 mL of ice water using a graduated cylinder and pour it into the beaker.
7. Using a weigh boat, weigh out 2 g of salt and add this to your beaker.
8. Use the thermometer to measure the temperature. Be sure to grasp the thermometer from the top and hold the thermometer in the middle of the solution (see picture on the left). DO NOT let the thermometer touch the bottom or the side of the beaker.

## Measuring Temperature

| Object | Estimated temperature ( $\left.{ }^{\circ} \mathrm{C}\right)$ | Measured temperature $\left({ }^{\circ} \mathrm{C}\right)$ |
| :--- | :--- | :--- |
| Tap water |  |  |
| Water with ice cubes and <br> 2.0 g of salt |  |  |

